

Ytterbium sesquiselenide Yb_2Se_3

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Ytterbium selenide, Yb_2Se_3 , was prepared by reacting the elements at 1173 K in an evacuated silica tube in an Sn flux. The ytterbium sesquiselenide crystallizes in the orthorhombic space group $Fddd$, adopting the Sc_2S_3 structure type. Its structure consists of edge-sharing (slightly distorted) YbSe_6 octahedra, and may be regarded as a defect NaCl structure.

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Key indicators

Single-crystal X-ray study

 $T = 298 \text{ K}$ Mean $\sigma(\text{Yb}-\text{Se}) = 0.001 \text{ \AA}$ R factor = 0.033 wR factor = 0.074

Data-to-parameter ratio = 45.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

The sesquichalcogenides of the rare earth elements adopt different structure types. While the La–Gd chalcogenides crystallize in defect variants of the Th_3P_4 type (Mauricot *et al.*, 1995), Yb_2S_3 adopts the $\alpha\text{-Al}_2\text{O}_3$ structure (El Fadli *et al.*, 1994). Based on powder diffractograms, the Sc, Y and Dy–Yb selenides (Dismukes & White, 1965; Flahaut *et al.*, 1965) were reported to form the Sc_2S_3 structure (Tremblet *et al.*, 1963). The cell dimensions for Yb_2Se_3 (orthorhombic system) were determined by Dismukes & White (1965) to be $a = 11.274 \text{ \AA}$, $b = 8.021 \text{ \AA}$, $c = 23.98 \text{ \AA}$ and $V = 2168.5 \text{ \AA}^3$, and by Flahaut *et al.* (1965) to be $a = 11.27 \text{ \AA}$, $b = 8.02 \text{ \AA}$, $c = 23.96 \text{ \AA}$ and $V = 2165.6 \text{ \AA}^3$. The atomic positions of Yb_2Se_3 were not refined in either case; Flahaut *et al.* (1965) merely extrapolated them from the ideal NaCl structure type. Our single-crystal structure study on Yb_2Se_3 confirms the suggested Sc_2S_3 type, and delivers crystallographic details with high precision. It is evident that the shifts from the NaCl structure are significant; *e.g.* the x parameters of Se1 and Se2 are not $3/8 = 0.375$, but 0.37062 (10) for Se1 and 0.38115 (7) for Se2. This is reflected in deviations from the ideal Se–Yb–Se bond angles of up to 4.5° and Yb–Se bond lengths varying from 2.7943 (6) to 2.8184 (6) \AA (Yb1) and from 2.7967 (6) to 2.8188 (3) \AA (Yb2).

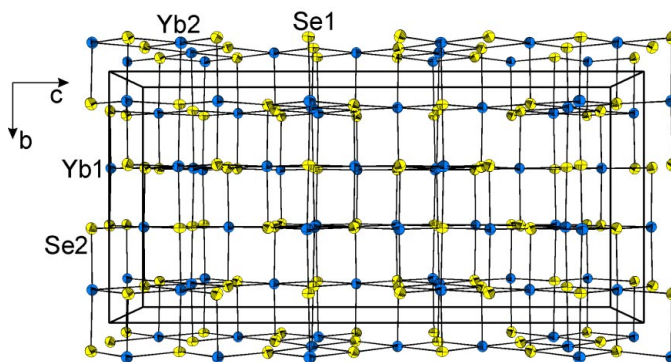


Figure 1

A projection of Yb_2Se_3 along the a axis. Displacement ellipsoids are drawn at the 99.9% probability level. Colour code: blue Yb and yellow Se.

Experimental

Yb₂Se₃ was obtained from a reaction of elemental ytterbium and selenium in a tin flux. The mixture was annealed at 1173 K over a period of 4 d, and then slowly cooled (3 K h⁻¹) to room temperature. Yb₂Se₃ crystallized in the form of black block-shaped crystals.

Crystal data

Yb₂Se₃
M_r = 582.96
 Orthorhombic, *Fddd*
a = 8.0183 (7) Å
b = 11.272 (1) Å
c = 23.961 (2) Å
V = 2165.7 (3) Å³
Z = 16
D_x = 7.152 Mg m⁻³

Mo *Kα* radiation
 Cell parameters from 5337 reflections
 θ = 3.2–35.0°
 μ = 54.32 mm⁻¹
T = 298 (2) K
 Block, black
 0.01 × 0.01 × 0.01 mm

Data collection

Bruker SMART APEX CCD diffractometer
 φ and ω scans
 Absorption correction: multi-scan (*SADABS*; Bruker, 1999)
T_{min} = 0.487, *T_{max}* = 0.581
 5337 measured reflections

1179 independent reflections
 937 reflections with *I* > 2σ(*I*)
R_{int} = 0.049
 θ_{\max} = 35.0°
h = -12 → 12
k = -17 → 17
l = -32 → 38

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.033
wR (*F*²) = 0.074
S = 1.12
 1179 reflections
 26 parameters

$w = 1/[\sigma^2(F_o^2) + (0.0244P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 (Δ/σ)_{max} = 0.001
 $\Delta\rho_{\max} = 1.70 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -3.52 \text{ e \AA}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.000397 (11)

Table 1

Selected geometric parameters (Å).

Yb1–Se2 ⁱ	2.7943 (6)	Yb2–Se2	2.7967 (6)
Yb1–Se2 ⁱⁱ	2.7943 (6)	Yb2–Se2 ⁱⁱⁱ	2.7967 (6)
Yb1–Se1	2.8095 (6)	Yb2–Se2 ^{vi}	2.8086 (7)
Yb1–Se1 ⁱⁱⁱ	2.8095 (6)	Yb2–Se2 ^{vii}	2.8086 (7)
Yb1–Se2 ^{iv}	2.8184 (6)	Yb2–Se1 ⁱ	2.8188 (3)
Yb1–Se2 ^v	2.8184 (6)	Yb2–Se1 ⁱⁱ	2.8188 (2)

Symmetry codes: (i) $\frac{1}{2} - x, -y, \frac{1}{2} - z$; (ii) $x - \frac{1}{4}, \frac{1}{4} + y, \frac{1}{2} - z$; (iii) $\frac{1}{4} - x, \frac{1}{4} - y, z$; (iv) $x - \frac{1}{2}, y, z - \frac{1}{2}$; (v) $\frac{3}{4} - x, \frac{1}{4} - y, z - \frac{1}{2}$; (vi) $x - \frac{1}{2}, \frac{1}{4} - y, \frac{3}{4} - z$; (vii) $\frac{3}{4} - x, y, \frac{3}{4} - z$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ATOMS* (Dowty, 1999); software used to prepare material for publication: *SHELXL97*.

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